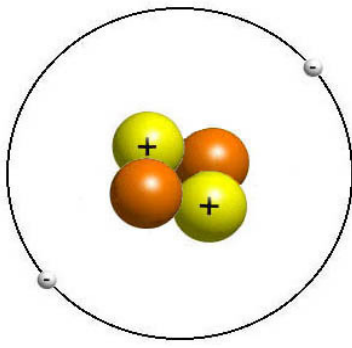
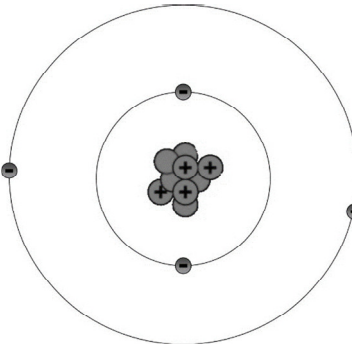
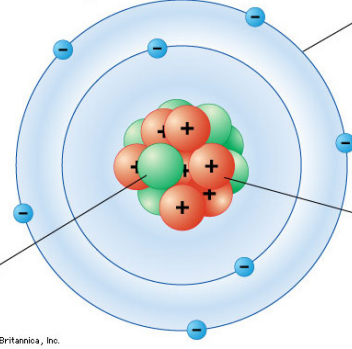
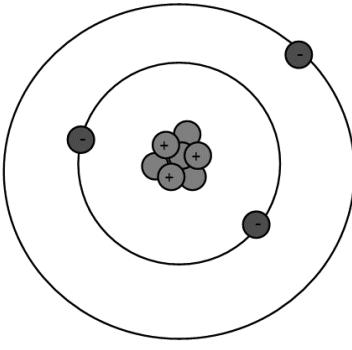
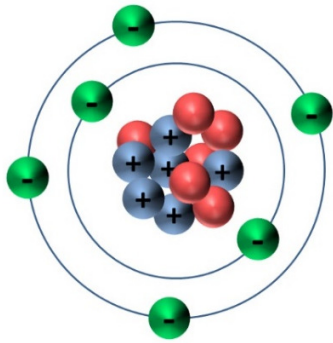


Name the Atom

Name the following elements and complete the information (use your periodic table)

	<p>Number of protons _____</p> <p>Number of electrons _____</p> <p>Number of Neutrons _____</p> <p>Number of valence electrons _____</p> <p>Atomic mass _____</p> <p>Name of the atom _____</p>
	<p>Number of protons _____</p> <p>Number of electrons _____</p> <p>Number of Neutrons _____</p> <p>Number of valence electrons _____</p> <p>Atomic mass _____</p> <p>Name of the atom _____</p>
	<p>Number of protons _____</p> <p>Number of electrons _____</p> <p>Number of Neutrons _____</p> <p>Number of valence electrons _____</p> <p>Atomic mass _____</p> <p>Name of the atom _____</p>
	<p>Number of protons _____</p> <p>Number of electrons _____</p> <p>Number of Neutrons _____</p> <p>Number of valence electrons _____</p> <p>Atomic mass _____</p> <p>Name of the atom _____</p>



Number of protons _____

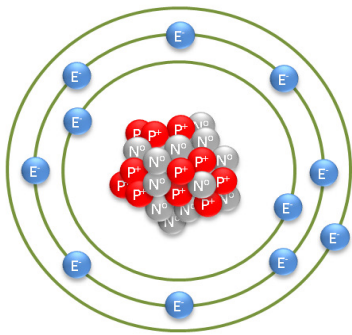
Number of electrons _____

Number of Neutrons _____

Number of valence electrons _____

Atomic mass _____

Name of the atom _____



Number of protons _____

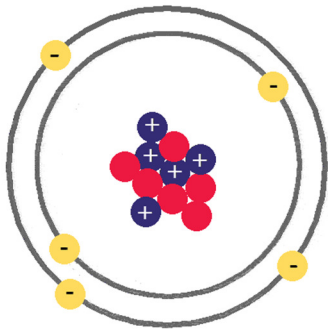
Number of electrons _____

Number of Neutrons _____

Number of valence electrons _____

Atomic mass _____

Name of the atom _____



Number of protons _____

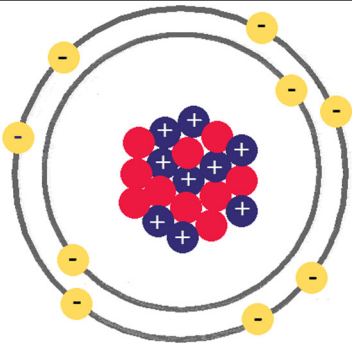
Number of electrons _____

Number of Neutrons _____

Number of valence electrons _____

Atomic mass _____

Name of the atom _____



Number of protons _____

Number of electrons _____

Number of Neutrons _____

Number of valence electrons _____

Atomic mass _____

Name of the atom _____

How were you able to determine the identity of the atom? Explain

How can you determine the mass of the atom?